

OL/2021(2022)/34/E-I

සියලු ම හිමිකම් ඇවිරිණි / முழுப் பதிப்புரிமையுடையது / All Rights Reserved

ශ්‍රී ලංකා විභාග දෙපාර්තමේන්තුව ශ්‍රී ලංකා විභාග දෙපාර්තමේන්තුව ශ්‍රී ලංකා විභාග දෙපාර්තමේන්තුව ශ්‍රී ලංකා විභාග දෙපාර්තමේන්තුව ශ්‍රී ලංකා විභාග දෙපාර්තමේන්තුව
 இலங்கைப் பரீட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம் இலங்கைப் பரීட்சைத் திணைக்களம்
 Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka Department of Examinations, Sri Lanka
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34 E I


අධ්‍යයන පොදු සහතික පත්‍ර (සාමාන්‍ය පෙළ) විභාගය, 2021(2022)
 கல்விப் பொதுத் தராதரப் பத்திர (சாதாரண தர)ப் பரீட்சை, 2021(2022)
 General Certificate of Education (Ord. Level) Examination, 2021(2022)

විද්‍යාව I
 விஞ்ஞானம் I
 Science I

පැය එකයි
 ஒரு மணித்தியாலம்
 One hour

Note :

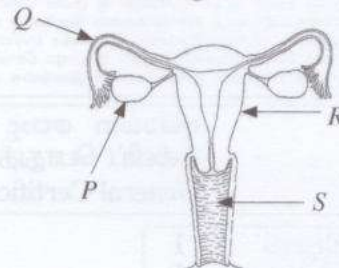
- * Answer all questions.
- * In each of the questions 1 to 40, pick one of the alternatives (1), (2), (3), (4) which you consider is correct or most appropriate.
- * Mark a cross (X) on the number corresponding to your choice in the answer sheet provided.
- * Further instructions are given on the back of the answer sheet. Follow them carefully.

1. To which of the following organizational level does the human heart belong?
 (1) cell (2) tissue (3) organ (4) system
 2. The hydrocarbons contained in LP gas as its major components are
 (1) methane and ethane. (2) propane and butane.
 (3) butane and pentane. (4) propane and pentane.
 3. What is the unit of the amount of work?
 (1) kg m s^{-1} (2) kg m s^{-2} (3) $\text{kg m}^{-1} \text{s}^{-1}$ (4) $\text{kg m}^2 \text{s}^{-2}$
 4. Which of the following is **not** a characteristic of the cardiac muscle tissue?
 (1) cells being mononuclear (2) occurrence of intercalated discs
 (3) cells being branched (4) acting voluntarily
 5. A seed observed by a student during a field study is shown in the diagram.
 This seed is dispersed by
 (1) animals. (2) water.
 (3) wind. (4) explosion.
- 
6. Which of the following quantity increases uniformly in an object moving with a constant acceleration?
 (1) distance (2) displacement (3) velocity (4) retardation
 7. Which of the following is equal in ${}^{40}_{19}\text{K}$ and ${}^{40}_{20}\text{Ca}$ atoms?
 (1) number of electrons
 (2) number of neutrons
 (3) sum of the number of electrons and protons
 (4) sum of the number of protons and neutrons
 8. Which property of water causes water to be used as a cooling agent to remove excessive heat generated in an automobile engine?
 (1) having a high specific heat capacity (2) having a high boiling point
 (3) being a colourless liquid (4) having a high density
 9. In the process of digestion of food, bile necessary for the emulsification of lipids is produced in the
 (1) gall bladder. (2) liver. (3) duodenum. (4) pancreas.
 10. What is the essential element for the production of thyroxine hormone in the thyroid gland?
 (1) sodium (2) phosphorus (3) calcium (4) iodine
 11. How much is the mass of NaCl contained in 100 cm^3 of a sodium chloride solution of concentration 1.0 mol dm^{-3} ? (Na = 23, Cl = 35.5)
 (1) 585 g (2) 58.5 g (3) 5.85 g (4) 0.585 g

- Questions 12 and 13 are based on the female reproductive system given in the diagram.

12. R and S respectively represent

- (1) uterus and vagina.
- (2) uterus and fallopian tube.
- (3) ovary and vagina.
- (4) ovary and fallopian tube.



13. An ovum fertilizes with a sperm in

- (1) P .
- (2) Q .
- (3) R .
- (4) S .

14. The image of a real object produced by a convex mirror is always

- (1) real and smaller than the object.
- (2) real and larger than the object.
- (3) virtual and smaller than the object.
- (4) virtual and larger than the object.

15. What is the increasing order of pH of four solutions of HCl, CH₃COOH, NaOH and NH₃ of equal concentration?

- (1) HCl < CH₃COOH < NaOH < NH₃
- (2) HCl < CH₃COOH < NH₃ < NaOH
- (3) CH₃COOH < HCl < NaOH < NH₃
- (4) CH₃COOH < HCl < NH₃ < NaOH

16. What is the number of CO₂ molecules in 22 g of carbon dioxide?

(C = 12, O = 16, Avagadro constant = 6.022 × 10²³ mol⁻¹)

- (1) $\frac{22}{44} \times 6.022 \times 10^{23}$
- (2) $\frac{44}{22} \times 6.022 \times 10^{23}$
- (3) $\frac{6.022 \times 10^{23}}{44 \times 22}$
- (4) $44 \times 22 \times 6.022 \times 10^{23}$

17. The chemical formula of the chemical fertilizer urea is CO(NH₂)₂. Select the correct statement about urea.

- (1) Urea possesses the properties of the constituent elements.
- (2) A molecule of urea has two carbon atoms.
- (3) A molecule of urea has four hydrogen atoms.
- (4) The total number of atoms in a urea molecule is seven.

18. Which option contains only the components that are found in the glomerular filtrate but not in urine of a healthy person?

- (1) glucose, amino acids
- (2) water, glucose
- (3) urea, amino acids
- (4) water, urea

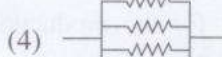
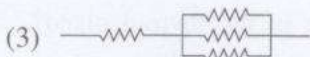
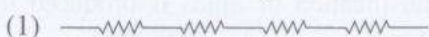
19. A jack fruit of mass 10 kg is falling freely from a height of 5 m of a jack tree. What is the velocity acquired by the fruit when it touches the ground? (Acceleration due to gravity is 10 m s⁻². Assume that the resistance of air is negligibly small.)

- (1) 5 m s⁻¹
- (2) 10 m s⁻¹
- (3) 50 m s⁻¹
- (4) 100 m s⁻¹

20. Select the correct statement regarding the circulation of blood in humans.

- (1) Always arteries contain oxygenated blood.
- (2) When circulated once through the body, blood goes twice through lungs.
- (3) Left ventricle acts as the pump of systemic blood circulation.
- (4) Always veins circulate blood away from the heart.

21. A few ways of connecting four resistors of equal magnitude are shown below. Of them, which one has the lowest equivalent resistance?



22. Consider the following organs.

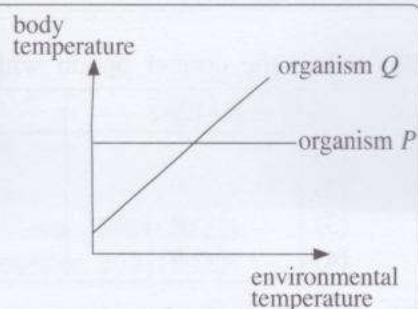
A - skin B - kidneys C - pancreas

Of these, those that contribute to maintain homeostasis of the body is/are

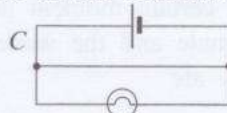
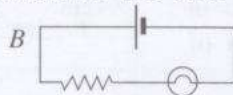
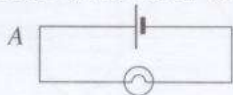
- (1) only A.
- (2) only A and B.
- (3) only B and C.
- (4) all A, B and C.

23. The graphs P and Q respectively indicate how the body temperature of the two organisms P and Q varies with the temperature of the environment. The organisms P and Q respectively can be

- (1) ox and bat.
- (2) fowl and tortoise.
- (3) elephant and duck.
- (4) frog and Thilapia.



24. In which order does the brightness of the bulb decrease in the circuits A , B and C ?



- (1) $A > B > C$
- (2) $A > C > B$
- (3) $B > C > A$
- (4) $C > A > B$

25. Some information regarding an element are as follows.

- Has several allotropic forms.
- One allotropic form conducts electricity.

This element is

- (1) carbon.
- (2) oxygen.
- (3) sulphur.
- (4) iron.

26. Some functions of the human brain are given below.

A - reception of senses

B - maintaining the balance of the body

C - controlling the rate of heart beat

Of the above, cerebellum controls

- (1) only A .
- (2) only B .
- (3) only A and B .
- (4) only B and C .

27. Consider the following statements.

A - The rate of a reaction increases when the temperature increases.

B - The kinetic energy of reactant particles increases when the temperature increases.

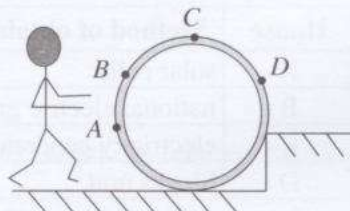
Of these

- (1) both statements A and B are true.
- (2) statement A is true and statement B is false.
- (3) both statements A and B are false.
- (4) statement A is false and statement B is true.

28. Which of the following phenomena **cannot** be explained by the Archimedes principle?

- (1) a ship cruising in the sea sinking more when entering a river
- (2) a balloon filled with helium gas moving upwards through air
- (3) floating of sunk orange seeds in a glass of orange juice when dissolving sugar
- (4) lifting a large mass by a hydraulic jack by applying a small force

29. The diagram shows a cross section of a cylindrical concrete ring kept on a flight of steps. Indicated as A , B , C and D are four points at which force can be applied by a man to roll it to the upper level. What is the point at which force can be applied on the cylinder in a suitable direction to roll it up with a lowest force?



- (1) A
- (2) B
- (3) C
- (4) D

30. When thin iron wool is heated, it instantly burns. But though an iron nail is heated till red hot, it does not undergo a considerable change. What conclusion can be drawn from these observations?

- (1) Iron wool acts as a catalyst.
- (2) Iron wool and iron nail undergo reactions different from each other.
- (3) The composition of iron wool and iron nail is different from each other.
- (4) The rate of a reaction depends on the physical nature of the reactants.

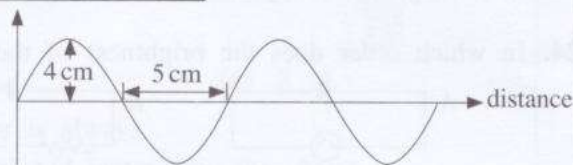
31. In 2011, the number of elephants in Sri Lanka was 5879. What is the most suitable biospherical organizational level to denominate that number of elephants?

- (1) species
- (2) population
- (3) community
- (4) eco-system

32. Select the correct option with regard to an *npn* transistor.

	Base	Emitter	Collector
(1)	<i>p</i>	<i>n</i>	<i>n</i>
(2)	<i>n</i>	<i>p</i>	<i>n</i>
(3)	<i>p</i>	<i>n</i>	<i>p</i>
(4)	<i>n</i>	<i>n</i>	<i>p</i>

33. The situation of transverse wave travelling on the surface of water at a certain moment is shown in the figure. The amplitude and the wave length of this wave respectively are

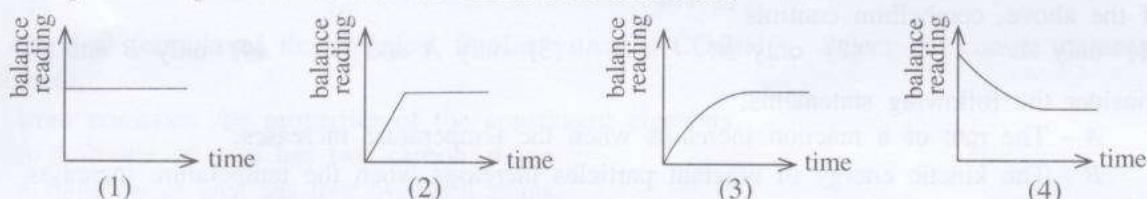


- (1) 4 cm and 5 cm. (2) 4 cm and 10 cm.
 (3) 5 cm and 4 cm. (4) 8 cm and 10 cm.

34. Which of the following statements is **false** about electroplating of iron?

- (1) An aqueous solution of a compound of the metal that is electroplated should be the electrolyte.
 (2) The electrolyte should be in high concentration for quality electroplating.
 (3) Iron should be made the cathode of the electrolytic cell.
 (4) The current used should be a direct current.

35. A conical flask containing aqueous hydrochloric acid is kept on a sensitive balance. Pieces of limestone were added to it in excess. Which of the following graphs indicates how the balance reading varies against time after the addition of limestone?



36. The number of turns in the primary coil and the secondary coil of a transformer are N_p and N_s respectively. The voltage difference of the primary coil is V_p and the voltage difference of the secondary coil is V_s . What is the correct relationship between the numbers of turns and the voltage differences?

- (1) $\frac{V_p}{V_s} = \frac{N_p}{N_s}$ (2) $\frac{V_s}{V_p} = \frac{N_p}{N_s}$ (3) $V_s V_p = N_s N_p$ (4) $V_s N_s = \frac{1}{V_p N_p}$

• Questions 37 to 40 are based on the information given in the following table. Information relating to the method of obtaining energy, lighting devices and cooking equipment used in four houses A, B, C and D are given below.

House	Method of obtaining energy	Lighting device	Cooking equipment
A	solar cells	CFL bulbs	LP gas cooker
B	national electric grid	filament bulbs	electric heater/cooker
C	electricity generator	fluorescent lamps	kerosene stove
D	biogas unit	biogas lamps	firewood hearth

37. For the energy obtaining method of which house is only a **non-renewable** energy source used?
 (1) A (2) B (3) C (4) D
38. Which houses use the lighting devices that largely contribute to pollute soil and water when improperly disposed to the environment?
 (1) A and C (2) A and D (3) B and C (4) B and D
39. Which house uses the cooking equipment that minimally contributes to pollute air inside the house?
 (1) A (2) B (3) C (4) D
40. What is the house that has taken most practical solutions to the electricity and fuel crisis?
 (1) A (2) B (3) C (4) D

අධ්‍යයන පොදු සහතික පත්‍ර (සාමාන්‍ය පෙළ) විභාගය, 2021(2022)
 கல்விப் பொதுத் தராதரப் பத்திர (சாதாரண தர)ப் பரீட்சை, 2021(2022)
 General Certificate of Education (Ord. Level) Examination, 2021(2022)

විද්‍යාව II
 விஞ்ஞானம் II
 Science II

පැය තුනයි
 மூன்று மணித்தியாலம்
 Three hours

අමතර කියවීමේ කාලය - මිනිත්තු 10 යි
 மேலதிக வாசிப்பு நேரம் - 10 நிமிடங்கள்
 Additional Reading Time - 10 minutes

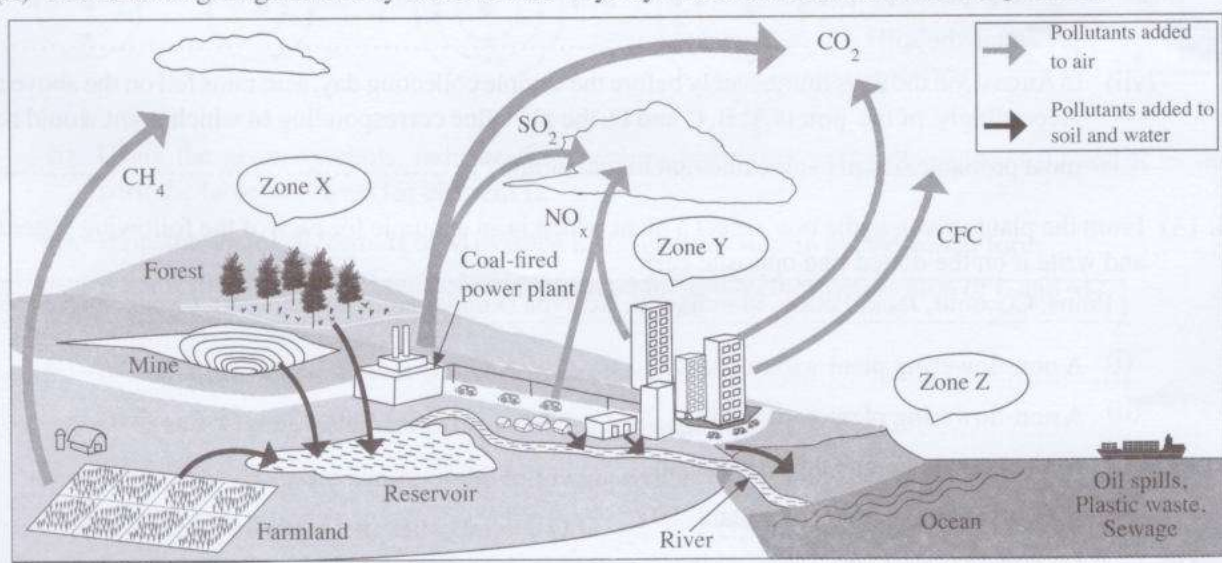
Use additional reading time to go through the question paper, select the questions you will answer and decide which of them you will prioritise.

Index Number:

- Instructions:**
- * Write your answers in neat handwriting.
 - * Answer the four questions in Part A, in the space provided.
 - * Of the five questions in Part B answer three questions only.
 - * After answering, tie Part A and the answer script of Part B together and hand over.

Part A

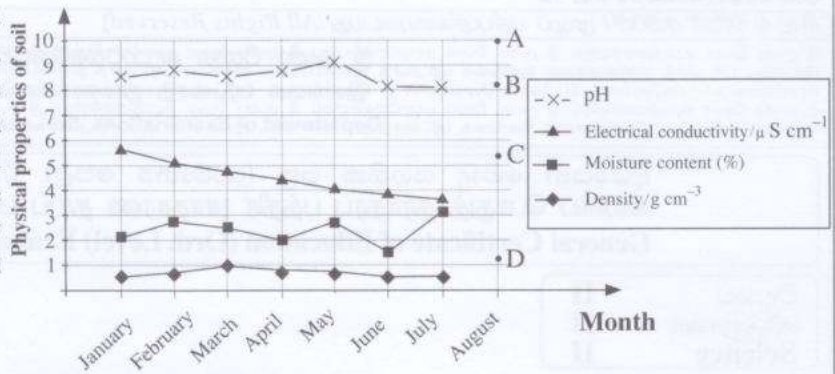
1. (A) The following diagram briefly indicates the ways by which air, soil and water are polluted.



Fill in the blanks in the table selecting an example from the diagram relevant to each of the following statements.

Statement	Example
(i) The gas contributing most to the increase in global warming
(ii) The main source that releases the components causing eutrophication in the reservoir
(iii) Gaseous organic compound that depletes the ozone layer
(iv) The zone most prone to have photochemical smog
(v) The source producing gases which cause acid rains
(vi) The source that adds ground heavy metals to topsoil
(vii) The pollutant accumulated in living organisms through food chains and subjected to the minimum chemical digestion in the digestive system

(B) In a certain area, a study was conducted about soil pollution. In this study, soil samples were collected monthly from a selected site and their physical properties density, moisture content, electrical conductivity and pH value were determined and plotted.



Answer the following questions based on the above graphs.

- (i) According to the density graph, what was the maximum density value reported?.....
- (ii) In which month was the moisture content at a minimum value?
- (iii) How much is the amount of moisture contained in 100 g of the soil sample collected in April?
- (iv) In which time range had the soil pH stayed constant?
- (v) Which physical property shows a continuous decrease during the period of data collection?
- (vi) Which physical properties shown in the graph are determined by the amount of ionic components present in the soil?
- (vii) In August, on the days immediately before the sample collecting day, acid rains fell on the above area. Accordingly, of the points A, B, C and D, the pH value corresponding to which point would be the most probable soil pH value relevant to that month?

2. (A) From the plants given in the box, select a plant which is an example for each of the following statements and write it on the dotted line opposite each.

Pinus, Coconut, Jack, Paddy, Marchantia, Acalypa (kuppamania/kuppameni), Cycas, Pogonatum

- (i) A non-flowering plant without seeds
- (ii) A non-flowering plant with seeds
- (iii) A monocotyledonous flowering plant
- (iv) A dicotyledonous flowering plant

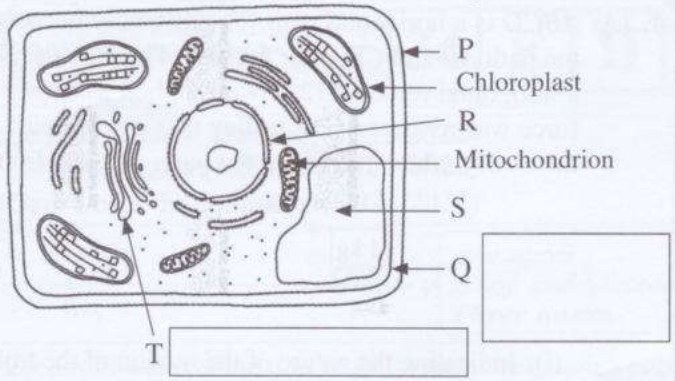
(B) In order to demonstrate the action of the enzyme amylase on starch during the process of digestion of food, a mixture was made by adding starch and amylase to water. The mixture was placed in a water bath at a temperature 37 °C. After five minutes, a drop from the mixture was taken out, a drop of iodine solution was added to it and the colour was observed. The above test was repeated at times given in the following table. The colour observed in each instance is shown in the table.

Time/minutes	5	15	25	35	45
Colour observed	Violet-blue	Blue	Blue	Yellow-brown	Yellow-brown

- (i) What is the compound formed by the action of amylase on starch in the aqueous medium?
- (ii) Give the reason for the following observations.
 - (a) Appearance of blue colour after 15 minutes
 - (b) Appearance of yellow-brown colour after 35 minutes
- (iii) Why is the mixture subjected to the experiment kept in a water bath at 37 °C?
- (iv) Which component does not undergo a chemical change though contributes to the chemical reaction pertaining to the above experiment?

(C) Given below is a diagram of a typical plant cell drawn based on the electron microscopic observations.

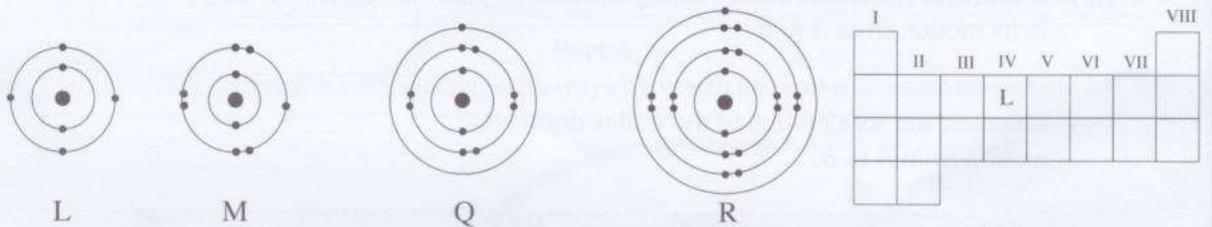
- (i) By which letter is the structure that helps to maintain the shape of plant cells named?
.....
- (ii) Write the names of the organelles labelled Q and T in the relevant boxes.
- (iii) By what letter is the organelle that can also be identified when observed under the optical microscope labelled?
.....



(iv) State the function carried out by the following organelles.

- (a) chloroplast
- (b) mitochondrion

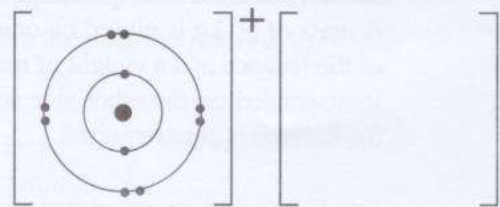
3. (A) How electrons exist in the energy levels belonging to atoms of elements L, M, Q and R are illustrated in the following diagrams. L, M, Q and R are not the standard symbols of those elements. Given on the right hand side is a periodic table indicating the places of the first twenty elements.



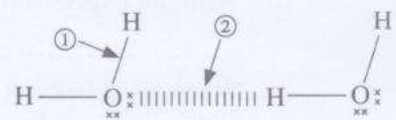
- (i) Using the given symbols, indicate the positions belonging to the elements M, Q and R in the periodic table as shown for element L.
- (ii) Write the chemical formula of M existing in the gaseous state in the molecular form.....
- (iii) What is the chemical formula of the compound formed by the combination of L and M?
.....
- (iv) Of L and M, which element is higher in electronegativity?
- (v) Of Q and R, which element is lower in first ionisation energy?
- (vi) State the acidic/basic nature of the following oxides formed by L and Q.



(vii) The diagram illustrates how Q exists in the ionic compound formed by the combination of Q and M. Draw how M exists in that compound.



(B) Chemical bonds within the water molecules and among the water molecules are indicated in the diagram by the arrows ① and ② respectively. Using it fill in the blanks in the following sentences.



- (i) The bond type indicated by the arrow ① is known as and the bond type indicated by the arrow ② is known as
- (ii) The type of bonds indicated by the arrow is responsible for the existence of relatively higher boiling point of water.
- (iii) A very small charge lies on the hydrogen atoms in the water molecules.

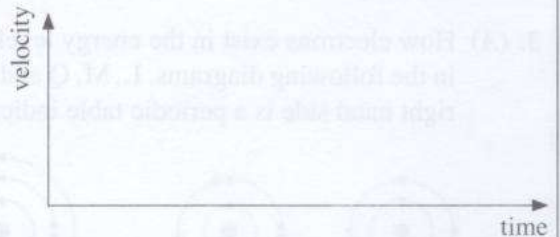
4. (A) *ABCD* is a horizontal path. The distances between the points *AB*, *BC* and *CD* are equal. The parts of the path *AB* and *CD* offer friction. The part *BC* is smooth. A trolley of mass 4 kg was placed at *A* and a horizontal force of 12 N was applied as shown in the figure. When the trolley reached *B*, the 12 N force was removed. The trolley that entered part *CD* came to rest at *D*. The frictional force acting on the trolley when moving in the parts offering friction was 4 N.



(i) Indicating the nature of the motion of the trolley from *A* to *D*, complete the following table.

	from <i>A</i> to <i>B</i>	from <i>B</i> to <i>C</i>	from <i>C</i> to <i>D</i>
Nature of motion of the trolley	acceleration

(ii) On the following system of axes, draw the approximate velocity-time graph for the motion of the trolley from *A* to *D*.



(iii) What is the unbalanced force acting on the trolley in its motion from *A* to *B*?

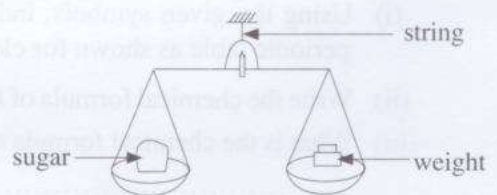
.....

(iv) Calculate the acceleration of the trolley during its motion from *A* to *B*.

.....

(B) The diagram shows a pan balance used in a retail shop.

(i) When 1 kg of sugar is placed on one pan of the balance, what is the mass of the weight that should be placed on the other pan to counterpoise the balance?

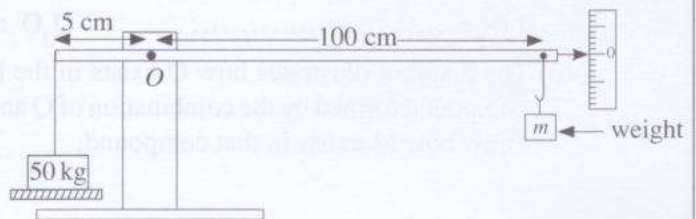


.....

(ii) What is the force acting on the string by which the balance is suspended when it is counterpoised as mentioned in (i) above? The mass of the balance only is 3 kg. ($g = 10 \text{ m s}^{-2}$)

.....

(iii) The figure shows a floor balance used in a wholesale store to weigh a large mass. A mass of 50 kg is placed on one side of the balance and a weight of mass *m* is suspended on the other side so that the balance is counterpoised.



(a) Find the moment created around point *O* by the mass of 50 kg.

(b) Write an expression in terms of *m* for the moment created around point *O* by the weight.

.....

(c) When the balance is counterpoised around point *O*, the anticlockwise moment created by the 50 kg mass and the clockwise moment created by the weight are equal. Find the value of *m*.

.....

(d) State **two** advantages of using a floor balance instead of a pan balance for measuring a large mass.

I.

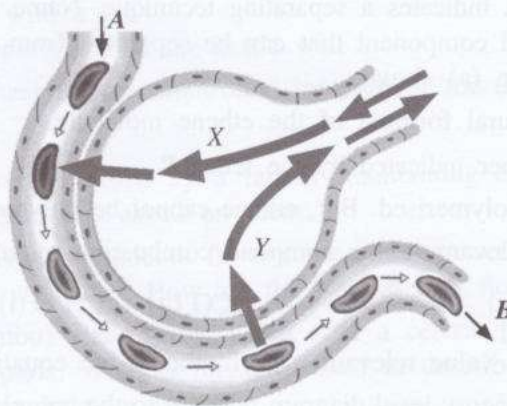
II.

Part B

- Answer only **three** questions from the questions No. 5, 6, 7, 8 and 9.

5. (A) Respiration is a biological process. The system organized for it is called the respiratory system.

- Write **two** changes occurring in inhaled air when it passes through the nasal cavity of man.
- Name the **two** structures which contain muscles contributing to increase the volume of the thoracic cavity during inspiration.
- A sketch of an alveolus in which gas exchange takes place in the respiration of man is given below. Indicated as X and Y in the figure are two types of gases that exchange in the alveolus.



- Name gas X and gas Y in respective order.
 - By what process does gas exchange occur between alveoli and blood capillaries?
 - What is the main difference between the blood that enters the capillary at A and the blood that leaves the capillary at B?
 - State an adaptation of alveoli to make the gas exchange efficient.
 - By what name is the ailing condition of gradual destruction of alveoli due to accumulation of compounds based on silica in alveoli known?
- (B) The process of producing food in green plants is called photosynthesis.
- Write the energy transformation occurring in the process of photosynthesis.
 - State how atmospheric carbon dioxide necessary for photosynthesis enters plant leaves.
 - "It is not possible to show water is an essential factor for photosynthesis by a simple laboratory experiment." Do you agree with this statement? Give a reason to validate your answer.
 - The water essential for photosynthesis is absorbed by roots and transported to plant leaves through the xylem tissue.
 - Of the cells that form the xylem tissue, what are the types of cells that contribute to transport water?
 - In addition to water, name another substance that is transported through the xylem tissue.
 - In addition to transport function, state the other function carried out by the xylem tissue.
 - Describe briefly how the cells in the xylem tissue are adapted to fulfil the function you stated in (c) above.

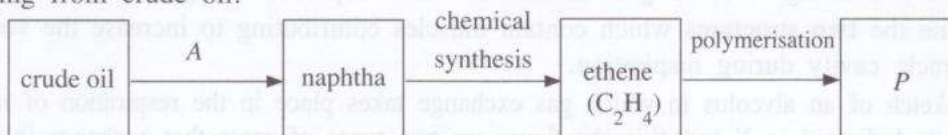
(20 marks)

6. (A) Sodium hydroxide (NaOH) is a basic chemical compound. It acts as a strong base in an aqueous solution.

- Explain what is a base according to the way it behaves in an aqueous solution.
- Why is sodium hydroxide called a strong base?
- Name an industrial use of sodium hydroxide.
- For a laboratory experiment, a student prepared 500 cm^3 of a sodium hydroxide solution with concentration of 1.00 mol dm^{-3} .
 - Name **two** laboratory glassware required to prepare the above solution.
 - How much is the mass of sodium hydroxide required to prepare the above solution? (H = 1, O = 16, Na = 23)

(c) It was experimentally found that the concentration of the solution prepared by the student was a little less than 1.00 mol dm^{-3} . State **two** errors that would have happened in the preparation of the solution causing that defect.

(B) The following flow chart indicates the manufacturing process of a commonly used polymer *P* starting from crude oil.



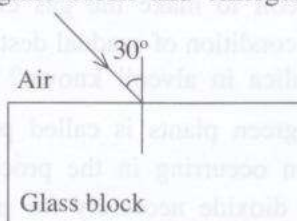
- (a) The letter *A* indicates a separating technique. Name that technique.
- (b) State a solid component that can be separated from crude oil using the technique you mentioned in (a) above.
- Draw the structural formula of the ethene molecule.
- Name the polymer indicated by the letter *P*.
- Ethene can be polymerised. But, ethane cannot be polymerised. What is the reason for this?
- The equation relevant to the complete combustion of ethene is given below.



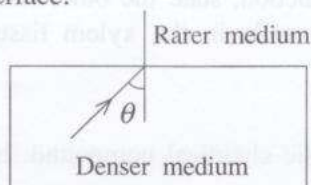
- What is the value relevant to *x* in the above equation?
- Sketch the energy level diagram relevant to the complete combustion of ethene indicating the levels where reactants and products exist.

(20 marks)

7. (A) The diagram shows how a ray of light coming through air falls on a block of glass. That ray of light travels through the glass block and emerges into the air again.



- Draw the rough sketch of the complete path of the light ray on your answer script.
- In the sketch, mark the angle of refraction '*r*' relevant to the first refraction of the ray.
- Write the equation which shows the relationship between the angle of incidence and angle of refraction. (Consider that the refractive index of glass with respect to air is *n*.)
- What is the value of the angle of emergence of the light ray?
- The following diagram indicates how a light ray travelling from a denser medium to a rarer medium falls on the interface.



- What is the path of the refracted ray when the angle of incidence θ in the denser medium is equal to the critical angle?
- By what name is the phenomenon happening known when the angle of incidence θ in the denser medium is greater than the critical angle?
- Write **two** examples for instances where the phenomenon in (b) above is made use of.

- (B) Water of mass 1 kg is contained in an electric kettle used in a house.
- What is the amount of heat required to raise the temperature of 1 kg of water from 20 °C to 100 °C? (**Specific heat capacity** of water is 4200 J kg⁻¹ °C⁻¹.)
 - Find the amount of heat absorbed by the kettle when the temperature of the water in the kettle was increased from 20 °C to 100 °C. (**Heat capacity** of the kettle is 160 J °C⁻¹.)
 - The power of the heating coil used to heat the kettle is 1000 W. What is the time taken to heat the water in the kettle from 20 °C to 100 °C?
 - The following measures have been taken to prevent the loss of heat from the kettle to the outer environment.
 - closing the kettle with a lid
 - polishing the outer surface of the kettle well

State the method of heat transfer controlled by each of the above measures. (20 marks)

8. (A) Some phenomena faced and identified by a farmer maintaining a mixed crop cultivation are given below. Answer the questions asked on them.
- The passion fruit cultivation produces flowers but does not set fruits. Therefore the flowers need to be artificially pollinated. How are the passion fruit flowers artificially pollinated?
 - Cocoyam (gahala/shembu) plants grow again after a certain period of time following the death of their aerial parts. What is the term used to describe this process by which the Cocoyam plants ensure their survival?
 - In the farmland, one bush of banana trees gives a greater yield. The plants in that bush are highly resistant to diseases. Name an artificial vegetative propagation method suitable to obtain a large number of banana plants with those traits at once.
 - The farmer intends to use a woodapple plant growing in the farmland as the stock and graft an orange twig to it. State **two** characteristics of the woodapple plant which would have caused its selection as the stock.
 - The farmer cultivated in the farmland a garden pea plant which was rare in the area where his farmland is situated. That garden pea plant bears round seeds. A majority of the new pea plants grown from that garden pea plant bears round seeds. But the rest bears wrinkled seeds. Based on your knowledge in genetics explain this phenomenon using the Punnett square.

- (B) In Figure 1, *AB* and *CD* indicate two metal rails. *PQ* is a conductor rod which can be slid on the two rails. The resistance of the rails and the rod are negligibly small. A uniform magnetic field is applied into the plane perpendicular to it on which the metal rails are placed. When the rod *PQ* is moved to right, an electric current is induced in it.

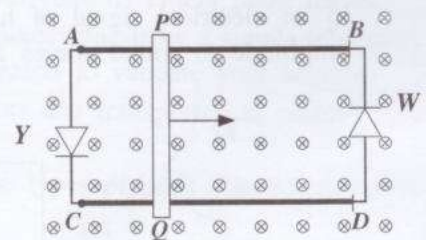


Figure 1

- Name the rule which can be used to decide the direction of the induced current in *PQ*.
- According to the rule stated in (i) above, does the current flow from *P* to *Q* or from *Q* to *P*?
- Due to the induced current in *PQ*, only one of the LEDs labelled *Y* and *W* lights.
 - What is the LED that lights?
 - State the reason why the other LED does not light.
- Figure 2 shows how a battery and a switch are connected between *A* and *C* after removing the two LEDs in Figure 1.



Figure 2

- State an observation made when the switch in the circuit is closed (put on).
- Name a device made by using the phenomenon associated with the observation made in (iv) (a) above.

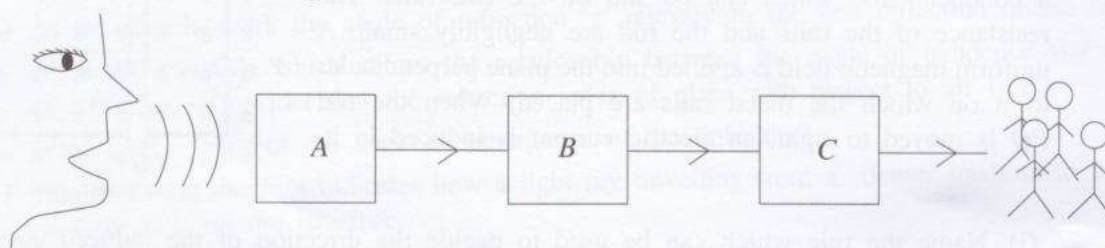
(20 marks)

9. (A) The following table gives some chemical properties of the metal copper (Cu) and two other metals X and Y. (X and Y are not their standard symbols.)

Metal	Reaction with water	Reaction with dilute hydrochloric acid
Cu	No reaction	No reaction
X	Reacts fast with cold water	Reacts violently
Y	No reaction with cold water but reacts with hot water	Reacts very fast

Answer the following questions using the above symbols where necessary.

- Write the above three metals in the decreasing order of their reactivity.
 - It is required to include hydrogen also in the series you made in (i) above according to its reactivity. Here, the place between which two metals does hydrogen occupy?
 - Of the metals given in the table, state the metal relevant to each of the following statements.
 - Stored in paraffin oil or kerosene without bringing into contact with air.
 - Extracted by reducing the metal oxide.
 - Y is a bivalent metal. Y reacts with dilute hydrochloric acid giving the metal chloride and a certain gas as products. Write the balanced chemical equation for this reaction.
 - A blue coloured aqueous solution of the sulphate of the metal copper was electrolysed using carbon electrodes.
 - Write the half reaction taking place on the cathode during electrolysis.
 - What kind of a colour change occurs in the electrolytic solution during the above electrolysis?
 - A simple cell is made using a copper metal plate and a zinc metal plate as electrodes and a dilute sulphuric acid solution as the electrolyte.
 - Which metal acts as the anode of the simple cell?
 - In which type of ion does the concentration remain constant in the solution of the electrolyte when the simple cell operates?
- (B) Equipment A converts the sound waves sent out by a speaker addressing a public meeting to an electrical signal of low amplitude. Equipment B converts the electrical signal of low amplitude to an electrical signal of high amplitude. Equipment C converts the electrical signal of high amplitude to sound waves again.



- Name the equipments A, B and C.
- Explain briefly the process taking place in equipment A.
- Draw the structure and the circuit symbol of the transistor, the main electronic device in equipment B and name its terminals.
- Name **two** main parts in the equipment C.
- In which characteristic of sound do the sound waves sent out by the speaker's mouth and the sound waves emitted by the equipment C differ from each other?
- When another speaker addresses the meeting in place of the first speaker, what characteristic of sound is different in the sound sent out by him?

(12) WWW.PastPapers.Wiki (12) (20 marks)

இலங்கைப் பரீட்சைத் திணைக்களம்
 இலங்கைப் பரீட்சைத் திணைக்களம்

රහස්‍ය

අ.පො.ස. (සා.පෙළ) විභාගය - 2021(2022)
 க.பொ.த. (சா.தர)ப் பரீட்சை - 2021(2022)

විෂය අංකය 34
 பாட இலக்கம்

විෂයය Science
 பாடம்

I පත්‍රය - පිළිතුරු
 I பத்திரம் - விடைகள்

ප්‍රශ්න අංකය வினா இல.	පිළිතුරේ අංකය விடை இல.	ප්‍රශ්න අංකය வினா இல.	පිළිතුරේ අංකය விடை இல.	ප්‍රශ්න අංකය வினா இல.	පිළිතුරේ අංකය விடை இல.	ප්‍රශ්න අංකය வினா இல.	පිළිතුරේ අංකය விடை இல.
01. 3	11. 3	21. 4	31. 2
02. 2	12. 1	22. 4	32. 1
03. 4	13. 2	23. 2	33. 2
04. 4	14. 3	24. 1	34. 2
05. 3	15. 2	25. 1	35. 4
06. 3	16. 1	26. 2	36. 1
07. 4	17. 3	27. 1	37. 3
08. 1	18. 1	28. 4	38. 1
09. 2	19. 2	29. 1/2	39. 2
10. 4	20. 3	30. 4	40. 4

විශේෂ උපදෙස් } එක් පිළිතුරකට ලකුණු
 විශේෂ අභිලාෂයක් } ඉටු සරියාන විධානවලට
 මුළු ලකුණු / මොත්තප්පු පුள்ளිකන් 01 x 40 = 40

පහත නිදසුනෙහි දක්වන පරිදි බහුවරණ උත්තරපත්‍රයේ අවසාන තීරුවේ ලකුණු ඇතුළත් කරන්න.
 කීழ් ප්‍රතිරෝධීය වීමට උපකාරී වන ජීවීන්ගේ ප්‍රජනන වේගය වැඩි කිරීමට උපකාරී වන ප්‍රතිජීවකාරක
 විනාශකරුවන්ගේ ප්‍රජනන වේගය වැඩි කිරීමට උපකාරී වන ප්‍රතිජීවකාරක

නිවැරදි පිළිතුරු සංඛ්‍යාව 25
 සරියාන විධානවලින් තොරව 40

I පත්‍රයේ මුළු ලකුණු 25
 පත්තிரம் I இன் மொத்தப்பள்ளி 40

සියලු අයිති ඇවැසි / முழுப் பதிப்புரிமைப்பது / All Rights Reserved

මහලු මාලු සංරක්ෂණය කිරීමේ මධ්‍යස්ථානයේ මුද්‍රණ කළ පොතකි. මෙහි ඇති අන්තර්ගතයන් ඔබගේ පුස්තකාලයේ ඇති පොතකට පිටුපත් කිරීමට ඔබට අවසරය නැත.
 Department of Examinations, Sri Lanka
 இலங்கைப் பரீட்சைத்துறை
 Department of Examinations, Sri Lanka

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අධ්‍යයන මහලු සහතික පත්‍ර (සාමාන්‍ය මට්ටම) විභාග, 2021(2022)
 கல்விப் பொதுத் தராதரப் பத்திர (சாதாரண தர)ப் பரீட்சை, 2021(2022)
 General Certificate of Education (Ord. Level) Examination, 2021(2022)

විද්‍යාව II
 விஞ்ஞானம் II
 Science II

පැය තුනයි
 முன்று மணிநேரம்
 Three hours

අමතර කියවීමේ කාලය - මිනිත්තු 10 යි
 மேலதிக வாசிப்பு நேரம் - 10 நிமிடங்கள்
 Additional Reading Time - 10 minutes

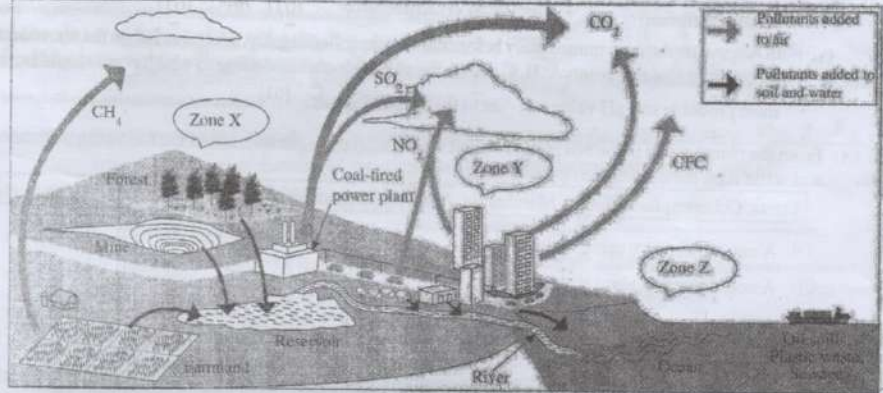
Use additional reading time to go through the question paper, select the questions you will answer and decide which of them you will prioritise.

Index Number: _____

Instructions: * Write your answers in neat handwriting.
 * Answer the four questions in Part A, in the space provided.
 * Of the five questions in Part B answer three questions only.
 * After answering, tie Part A and the answer script of Part B together and hand over.

Part A

1. (A) The following diagram briefly indicates the ways by which air, soil and water are polluted.

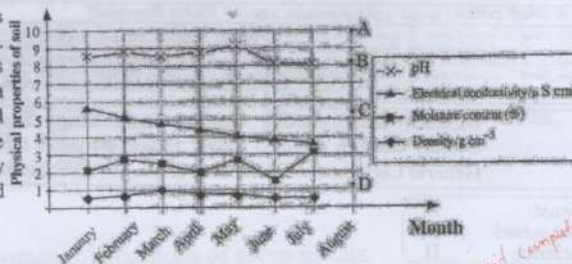


Fill in the blanks in the table selecting an example from the diagram relevant to each of the following statements.

Statement	Example
(i) The gas contributing most to the increase in global warming	CO ₂ / Carbon dioxide (01)
(ii) The main source that releases the components causing eutrophication in the reservoir	Farmland (01)
(iii) Gaseous organic compound that depletes the ozone layer	CFC / chlorofluorocarbon (01)
(iv) The zone most prone to have photochemical smog	Zone Y (01)
(v) The source producing gases which cause acid rains	Coal - fired power plant/ Coal/ Vehicles/Ships /Factories (01)
(vi) The source that adds ground heavy metals to topsoil	Mine (01)
(vii) The pollutant accumulated in living organisms through food chains and subjected to the minimum chemical digestion in the digestive system	Plastic (waste) (01)

07

(B) In a certain area, a study was conducted about soil pollution. In this study, soil samples were collected monthly from a selected site and their physical properties density, moisture content, electrical conductivity and pH value were determined and plotted.



Answer the following questions based on the above graphs.

- (i) According to the density graph, what was the maximum density value reported? 1 g cm⁻³ (01)
- (ii) In which month was the moisture content at a minimum value? June (01)
- (iii) How much is the amount of moisture contained in 100 g of the soil sample collected in April? 2 g / 2 / 2% (01)
- (iv) In which time range had the soil pH stayed constant? From June to July / June - July (01)
- (v) Which physical property shows a continuous decrease during the period of data collection? Electrical conductivity (01)
- (vi) Which physical properties shown in the graph are determined by the amount of ionic components present in the soil? Electrical conductivity (01) pH (01)
- (vii) In August, on the days immediately before the sample collecting day, acid rains fell on the above area. Accordingly, of the points A, B, C and D, the pH value corresponding to which point would be the most probable soil pH value relevant to that month? C (01)

2. (A) From the plants given in the box, select a plant which is an example for each of the following statements and write it on the dotted line opposite each.

Pinus, Coconut, Jack, Paddy, Marchantia, Acalypha (kuppamania/kuppamēni), Cycas, Pogonatum

- (i) A non-flowering plant without seeds Marchantia / Pogonatum (01)
- (ii) A non-flowering plant with seeds Cycas / Pinus (01)
- (iii) A monocotyledonous flowering plant Coconut / Paddy (01)
- (iv) A dicotyledonous flowering plant Jack / Acalypha / Kuppamania (01)

(B) In order to demonstrate the action of the enzyme amylase on starch during the process of digestion of food, a mixture was made by adding starch and amylase to water. The mixture was placed in a water bath at a temperature 37 °C. After five minutes, a drop from the mixture was taken out, a drop of iodine solution was added to it and the colour was observed. The above test was repeated at times given in the following table. The colour observed in each instance is shown in the table.

Time/minutes	5	15	25	35	45
Colour observed	Violet-blue	Blue	Blue	Yellow-brown	Yellow-brown

- (i) What is the compound formed by the action of amylase on starch in the aqueous medium? Maltose (01)
- (ii) Give the reason for the following observations. (01)
 - (a) Appearance of blue colour after 15 minutes Presence of starch (01)
 - (b) Appearance of yellow-brown colour after 35 minutes All starch are reacted (01)
- (iii) Why is the mixture subjected to the experiment kept in a water bath at 37 °C? Providing the optimal temperature for enzymatic action / Maintaining at the body temperature / To increase the rate of the reaction (01)
- (iv) Which component does not undergo a chemical change though contributes to the chemical reaction pertaining to the above experiment? Amylase / Enzyme (01)

(C) Given below is a diagram of a typical plant cell drawn based on the electron microscopic observations.

(i) By which letter is the structure that helps to maintain the shape of plant cells named?

P (01)

(ii) Write the names of the organelles labelled Q and T in the relevant boxes.

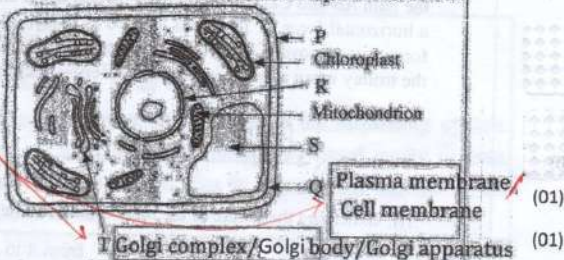
(iii) By what letter is the organelle that can also be identified when observed under the optical microscope labelled?

R / S (01)

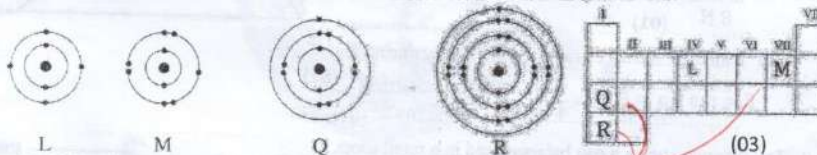
(iv) State the function carried out by the following organelles.

(a) chloroplast Carry out photosynthesis / Production of food (01)

(b) mitochondrion Cellular respiration / Aerobic respiration / Production of energy (01)



3. (A) How electrons exist in the energy levels belonging to atoms of elements L, M, Q and R are illustrated in the following diagrams. L, M, Q and R are not the standard symbols of those elements. Given on the right hand side is a periodic table indicating the places of the first twenty elements.



(i) Using the given symbols, indicate the positions belonging to the elements M, Q and R in the periodic table as shown for element L.

(ii) Write the chemical formula of M existing in the gaseous state in the molecular form. M_2 / F_2 (01)

(iii) What is the chemical formula of the compound formed by the combination of L and M? LM_4 / CF_4 (01)

(iv) Of L and M, which element is higher in electronegativity? M / F (01)

(v) Of Q and R, which element is lower in first ionisation energy? R / K (01)

(vi) State the acidic/basic nature of the following oxides formed by L and Q.

LO_2 : acidic (01) Q_2O : basic (01)

(vii) The diagram illustrates how Q exists in the ionic compound formed by the combination of Q and M. Draw how M exists in that compound.

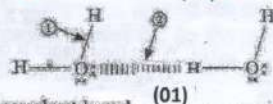


(B) Chemical bonds within the water molecules and among the water molecules are indicated in the diagram by the arrows ① and ② respectively. Using it fill in the blanks in the following sentences.

(i) The bond type indicated by the arrow ① is known as covalent bond and the bond type indicated by the arrow ② is known as inter molecular bond / inter molecular attraction (01)

(ii) The type of bonds indicated by the arrow ② is responsible for the existence of relatively higher boiling point of water. Hydrogen bond / inter - molecular forces (01)

(iii) A very small positive / + / δ^+ charge lies on the hydrogen atoms in the water molecules. (01)



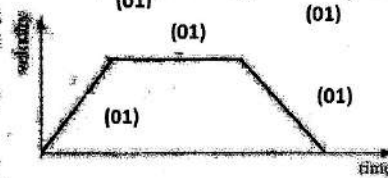
4. (A) ABCD is a horizontal path. The distances between the points AB, BC and CD are equal. The parts of the path AB and CD offer friction. The part BC is smooth. A trolley of mass 4 kg was placed at A and a horizontal force of 12 N was applied as shown in the figure. When the trolley reached B, the 12 N force was removed. The trolley that entered part CD came to rest at D. The frictional force acting on the trolley when moving in the parts offering friction was 4 N.



- (i) Indicating the nature of the motion of the trolley from A to D, complete the following table.

	from A to B	from B to C	from C to D
Nature of motion of the trolley	acceleration	uniform velocity	deceleration / retardation / negative acceleration

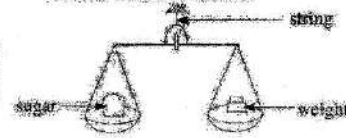
- (ii) On the following system of axes, draw the approximate velocity-time graph for the motion of the trolley from A to D.
- (iii) What is the unbalanced force acting on the trolley in its motion from A to B?



- (iv) Calculate the acceleration of the trolley during its motion from A to B.
- $F = ma / a = \frac{F}{m} = \frac{8 \text{ N}}{4 \text{ kg}} = 2 \text{ ms}^{-2}$ Any substitution (01)

- (B) The diagram shows a pan balance used in a retail shop.

- (i) When 1 kg of sugar is placed on one pan of the balance, what is the mass of the weight that should be placed on the other pan to counterpoise the balance?



- (ii) What is the force acting on the string by which the balance is suspended when it is counterpoised as mentioned in (i) above? The mass of the balance only is 3 kg. ($g = 10 \text{ ms}^{-2}$)

- (iii) The figure shows a floor balance used in a wholesale store to weigh a large mass. A mass of 50 kg is placed on one side of the balance and a weight of mass m is suspended on the other side so that the balance is counterpoised.

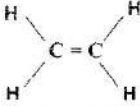
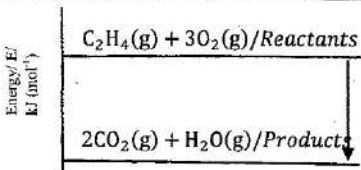


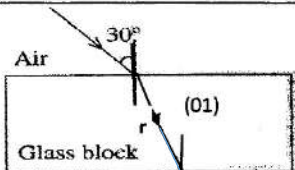
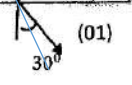
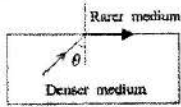
- (a) Find the moment created around point O by the mass of 50 kg. $500 \text{ N} \times \frac{5}{100} \text{ m} / 25 \text{ Nm} / 500 \text{ N} \times 5 \text{ cm} / 2500 \text{ Ncm}$ (01)
- (b) Write an expression in terms of m for the moment created around point O by the weight. $mg \times 1 \text{ m} / mg \times 100 \text{ cm} / 10m$ (01)
- (c) When the balance is counterpoised around point O, the anticlockwise moment created by the 50 kg mass and the clockwise moment created by the weight are equal. Find the value of m . $m = 2.5 \text{ kg}$ (01)
- (d) State two advantages of using a floor balance instead of a pan balance for measuring a large mass.
- I. Easy to handle (the weight) / Can measure a large mass
- II. No need of large standard weight to weigh / no need to lift the balance

Any two answers (02)

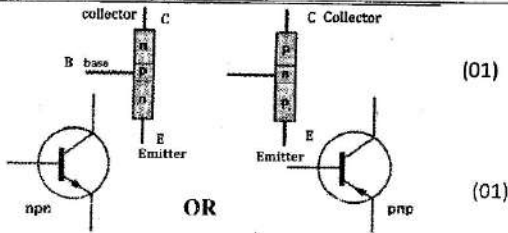
(See page five)

5	(A)	(i)	<ul style="list-style-type: none"> Moisturizing <u>or</u> humidifying inhaled air Bringing inhaled air to the body temperature Removal of impurities or foreign bodies from inhaled air <p>For any two</p>	2
		(ii)	<ul style="list-style-type: none"> Diaphragm (01) Ribs / Give marks even for inter - costal muscles. (01) 	2
		(iii) (a)	<ul style="list-style-type: none"> X - O₂ / Oxygen (01) Y - CO₂ / Carbondioxide <p>Give marks even for H₂O <u>or</u> water vapour (01)</p>	2
		(b)	Diffusion	1
		(c)	<p>A - oxygen concentration is low B - oxygen concentration is high A deoxygenated blood B - deoxygenated blood A - CO₂/HCO₃⁻ concentration is high B - CO₂/HCO₃⁻ concentration is low</p> <p>For any one</p>	1
		(d)	<ul style="list-style-type: none"> Alveolar walls being thin Alveolar walls being wet Alveolar walls being permeable Presence of a blood capillary network around alveoli Presence of a large number of air sacs <p>For any one</p>	1
		(e)	Silicosis	1
				10
	(B)	(i)	Light energy → Chemical energy	1
		(ii)	<u>Diffuses</u> (01) into leaves through <u>stomata</u> (01)	2
		(iii)	Yes (01) Because without water the plant will die / Cannot set up a control experiment without water (01)	2
		(iv) (a)	<ul style="list-style-type: none"> Xylem vessels (01) Tracheid (01) 	2
		(b)	Minerals	1
		(c)	Providing mechanical strength / support/ rigidity to the plant	1
		(d)	Lignification (of cell wall) <u>or</u> (Cell walls) are lignified.	1
			TOTAL MARKS	20

6.	(A)	(i)	Compounds that increase the OH^- ion concentration (in an aqueous solution) or compounds that release OH^- ions (to an aqueous solution)	1
		(ii)	Because NaOH completely dissociates/ionizes (in an aqueous solution) or $\text{NaOH(aq)} \rightarrow \text{Na}^+(\text{aq}) + \text{OH}^-(\text{aq})$ Physical status are not essential.	1
		(iii)	In the production of soap/ paper/ artificial silk/ dyes/ medicine or refining petroleum products For any one	1
		(iv)	(a) <ul style="list-style-type: none"> • Volumetric flask • Watch glass • Funnel For any two	2
		(b)	Molar mass of NaOH = $23 + 16 + 1 = 40 \text{ (g mol}^{-1}\text{)}$ (01) Number of moles of NaOH = $\frac{1}{1000} \times 500$ / $\frac{100}{1000} \times 500$ (01) Mass of NaOH = $0.5 \times 40 = 20 \text{ g}$ (01)	3
		(c)	<ul style="list-style-type: none"> • Weighing NaOH less than the amount required • Not transferring all NaOH to the flask • Overlooking errors in the balance • Not weighing accurately • Addition of water in excess • Not using distilled water For any two	2
				10
	B	(i)	(a) Fractional distillation	1
		(b)	<ul style="list-style-type: none"> • Wax (paraffin) • Bitumen / Tar / Asphalt • Grease For any one	1
		(ii)	 <p>For this $\text{CH}_2 = \text{CH}_2$ (01) only</p>	2
		(iii)	Polythene/ polyethylene/ polyethene	1
		(iv)	<ul style="list-style-type: none"> • Absence of double bond in ethane • Presence of only single bonds in ethane • Ethane is an alkane. • Presence of a double bond in ethene • Ethene is an alkene For any one	1
		(v)	(a) $x = 2$ or $\text{C}_2\text{H}_4(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow 2\text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{l}) + (\text{Heat})$	1
		(b)	 <p>Showing reactants and products (01) Labelling the axis (01) For the downward arrow (01)</p>	3
				10
			TOTAL MARKS	20

7	(A)	(i)		2
		(ii)	 <p>Showing 'r' in the diagram</p>	1
		(iii)	$n = \frac{\sin i}{\sin r} \quad \text{or} \quad n = \frac{\sin 30^\circ}{\sin r} \quad \text{or} \quad \sin r = \sin 30^\circ / n \quad \text{or} \quad \sin r = \frac{1}{2n}$ <p>or</p> $n = \frac{\sin \text{value of angle of incidence}}{\sin \text{value of angle of refraction}}$	2
		(iv)	30° or showing 30° in the diagram	1
		(v) (a)	 <p>or travels along the interface</p>	1
		(b)	Total internal reflection	1
		(c)	<ul style="list-style-type: none"> • Optic fibers • Internet connections • Periscope • Endoscope • Telephone communication technology • Decorations • Binocular <p style="text-align: right;">For any two</p>	2
				10
	(B)	(i)	$Q = m c \theta$ $= 1 \text{ (kg)} \times 4200 \text{ (J kg}^{-1} \text{ }^\circ\text{C}^{-1}) \times 80 \text{ (}^\circ\text{C)}$ $= 336\,000 \text{ J OR } 336 \text{ kJ}$ <p>For the equation (01) For the substitution (01) Answer with unit (01)</p>	3
		(ii)	$Q = C \theta \quad \text{or}$ $= 160 \text{ (J }^\circ\text{C}^{-1}) \times 80 \text{ (}^\circ\text{C)} \quad (01)$ $= 12\,800 \text{ J or } 12.8 \text{ kJ} \quad (01)$	2
		(iii)	$E = P t \quad \text{or} \quad t = E / P \quad \text{or}$ <p>For Calculating total energy (01) mark</p> $t = \frac{(336\,000 + 12\,800) \text{ or } 348\,800}{1000}$ <p>For the equation or substitution (01)</p> $t = 348.8 \text{ seconds or } 5.81 \text{ minutes} \quad \text{For the answer with unit (01)}$	3
		(iv) (a)	Convection	1
		(b)	Radiation	1
				10
			TOTAL MARKS	20

8.	(A)	(i)	Depositing the pollen of a flower artificially on the stigma of the same flower or on the stigma of a different flower of the same species with fingers or with a brush. Free mark *	1									
		(ii)	Survival in unfavourable time/ Perennation/ Dormant period / Dormancy	2									
		(iii)	Tissue culture	2									
		(iv)	<ul style="list-style-type: none"> • Having a strong root system • Being resistant to diseases and pests • Withstanding unfavourable environmental conditions • Having uniform growth • Both woodapple and orange are plants showing affinities <p style="text-align: right;">For any two</p>	2									
		(v)	<p>Any letter can be used instead of R (r) Gametes</p> <table border="1" style="display: inline-table; margin-right: 10px;"> <tr> <td></td> <td>R</td> <td>r</td> </tr> <tr> <td>R</td> <td>RR</td> <td>Rr</td> </tr> <tr> <td>r</td> <td>Rr</td> <td>rr</td> </tr> </table> <p>For showing genotypes (01) (01) (for the punnett square)</p> <p>Indicating the ratio between round seeds and wrinkled seeds as 3 : 1 (01) Deduct 1 mark for any other explanation without the Punnett square.</p>		R	r	R	RR	Rr	r	Rr	rr	3
	R	r											
R	RR	Rr											
r	Rr	rr											
				10									
(B)	(i)		Fleming's right-hand rule	2									
	(ii)		Q to P	1									
	(iii)	(a)	Y	1									
		(b)	Because LED / W is reverse biased Give only 1 mark for stating LED/W dose not allow the current to flow	2									
	(iv)	(a)	(PQ conductor rod) moves towards the right. For stating that 'PQ moves' give only 1 mark	2									
		(b)	(DC) motor or speaker (loudspeaker)/Galvanometer/Ammeter/Voltmeter	2									
				10									
			TOTAL MARKS	20									

9	(A)	(i)	X, Y, Cu <u>or</u> X > Y > Cu	1	
		(ii)	In between Y and Cu	1	
		(iii)	(a)	X	1
			(b)	Cu	1
		(iv)	Mg $Y + 2HCl \rightarrow YCl_2 + H_2$ For correct reactants and products (01) For the balanced equation (01) If Mg is used instead of Y marks will be awarded.	2	
		(v)	(a)	$Cu^{2+}(aq) + 2e \rightarrow Cu(s)$ Award marks even without physical states	1
	(b)		(Intensity of) blue color decreases / becomes colorless	1	
	(vi)	(a)	Zn / Zinc plate	1	
		(b)	SO_4^{2-} / Sulphate ion	1	
	10				
	(B)	(i)	A – Microphone (01) B – Amplifier (01) C – Loudspeaker / speaker (01)	3	
		(ii)	Sound waves vibrate the diaphragm and induce an electromotive force / a current.	1	
		(iii)	 <p style="text-align: center;">OR</p> <p style="text-align: center;">It is compulsory to label terminals either in the structure or the symbol.</p>	2	
(iv)	<ul style="list-style-type: none"> • Coil • Magnet • Paper / cardboard cone / <i>Diaphragm</i> <p style="text-align: right;">For any two</p>	2			
(v)	Loudness	1			
(vi)	Pitch / Loudness / Quality of sound	1			
10					
20					

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